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N,*N*-Diethyl-4-(methylsulfonyl)-2-nitroaniline

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Abstract

The title compound, $C_{11}H_{16}N_2O_4S$, shows significant rotation of the NEt₂ and NO₂ groups from the benzene plane, which reduces the steric compression of these *ortho* substituents. The methyl groups of the NEt₂ substituent adopt a *syn* arrangement, unlike the *anti* arrangement in N_1N_2 -diethyl-2,4-dinitroaniline.

Comment

The methyl groups in the NEt₂ unit of the title compound (1) adopt a syn arrangement. The NEt₂ and NO_2 N atoms are displaced by -0.139(3) and 0.283 (3) Å from the mean plane of the benzene ring whilst the S atom lies 0.008(1) Å from it. The orientations of the NEt2 and NO2 groups are described, respectively, by the torsion angles C5—C4—N2—C10 22.6 (4), C3—C4—N2—C8 31.0 (4), C4—C3—N1— O3 38.1 (4), C2—C3—N1—O4 40.6 (4)°. The NEt₂ N atom is slightly pyramidal, N2 being displaced from the plane of the directly attached C atoms by 0.070 (3) Å. The nitro group is almost planar with the N atom 0.030(3) Å from the O3-O4-C3 plane. The differences between the two N-C(ethyl) and between the two C-C(ethyl) bond lengths are probably due to librational effects. One large anisotropic displacement parameter for C11 suggests either a large component of vibration amplitude or slight disorder.

© 1995 International Union of Crystallography Printed in Great Britain – all rights reserved The *syn* arrangement of the methyl groups contrasts with an *anti* arrangement in *N*,*N*-diethyl-2,4-dinitro-aniline (MacNicol & Mallinson, unpublished results). The N2—C4 and N1—C3 distances in the title compound are 1.366(5) and 1.462(5) Å compared with 1.353(5) and 1.458(5) Å in the dinitro compound. The significantly closed angle C3—C4—C5 here, 114.0(4)°, may be compared with 115.2(4)° in the dinitro compound.

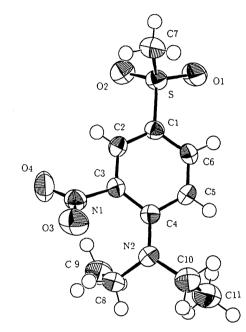


Fig. 1. A view of the molecular structure showing the atom labels. Displacement ellipsoids are shown at the 50% probability level; H atoms are shown as circles of arbitrary radii and are labelled according to the atoms to which they are attached.

Experimental

Crystals, m.p. 359 K, were prepared at Glasgow University by Dr C. W. N. Horning.

Crystal data

C11H16N2O4S Mo $K\alpha$ radiation $\lambda = 0.71069 \text{ Å}$ $M_r = 272.32$ Cell parameters from 25 Monoclinic reflections $P2_1/n$ $\theta = 11 - 15^{\circ}$ a = 5.035(1) Å $\mu = 0.249 \text{ mm}^{-1}$ b = 11.684(1) ÅT = 293 Kc = 21.792(3) ÅNeedle $\beta = 90.54(2)^{\circ}$ $0.8 \times 0.1 \times 0.1 \text{ mm}$ $V = 1282 (1) \text{ Å}^3$ Yellow $D_x = 1.41 \text{ Mg m}^{-3}$

Data collection

Enraf-Nonius CAD-4 diffractometer

 $R_{\text{int}} = 0.023$ $\theta_{\text{max}} = 27^{\circ}$

Acta Crystallographica Section C ISSN 0108-2701 ©1995 ω -2 θ scans $h=0 \rightarrow 6$ Absorption correction: $k=0 \rightarrow 14$ $l=-27 \rightarrow 27$ 2865 measured reflections 2797 independent reflections 1386 observed reflections $[I>2\sigma(I)]$ intensity decay: none

Refinement

Refinement on F	$(\Delta/\sigma)_{\text{max}} = 0.23$
R = 0.044	$\Delta \rho_{\text{max}} = 0.6 \text{ e Å}^{-3}$
wR = 0.053	$\Delta \rho_{\min} = -0.4 \text{ e Å}^{-3}$
S = 2.3	Extinction correction: none
1386 reflections	Atomic scattering factors
223 parameters	from International Tables
All H-atom parameters	for X-ray Crystallography
refined except H11B	(1974, Vol. IV, Tables
which was not refined	2.2B and 2.3.1)
$w = 1/\sigma^2(F)$	

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å²)

 $U_{eo} = (1/3)\sum_{i}\sum_{i}U_{ii}a^{*}a^{*}\mathbf{a}_{i}.\mathbf{a}_{i}.$

$Oeq = (1/3) \square_i \square_j Oij a_i a_j a_i a_j$						
	x	у	z	$U_{ m eq}$		
C1	0.6510(6)	0.2527 (3)	0.5213(2)	0.04 i		
C2	0.7487 (7)	0.3245 (3)	0.5662(2)	0.043		
C3	0.9105 (6)	0.2825(3)	0.6118 (2)	0.040		
C4	0.9737 (7)	0.1654(3)	0.6179(2)	0.047		
C5	0.8763 (9)	0.0965 (4)	0.5695 (2)	0.057		
C6	0.7225 (8)	0.1383(3)	0.5228(2)	0.051		
C 7	0.6773 (10)	0.3801 (5)	0.4144 (2)	0.059		
C8	1.1001 (8)	0.1639 (4)	0.7289(2)	0.058		
C9	0.8259 (9)	0.1867 (6)	0.7510(2)	0.075		
C10	1.2456 (11)	0.0061 (5)	0.6587(3)	0.079		
C11	1.0771 (17)	-0.0869 (5)	0.6793(3)	0.110		
N1	1.0388 (6)	0.3703(3)	0.6496(1)	0.052		
N2	1.1107 (7)	0.1187(3)	0.6661 (2)	0.061		
O1	0.3417 (5)	0.2130(3)	0.4285(1)	0.073		
O2	0.2740 (5)	0.3922(2)	0.4863(1)	0.063		
O3	1.2762 (5)	0.3600(3)	0.6618(1)	0.073		
O4	0.9096 (6)	0.4537 (3)	0.6643(1)	0.077		
S	0.45238 (17)	0.30780 (9)	0.46148 (4)	0.049		

Table 2. Selected geometric parameters (Å, °)

C1—C2	1.377 (5)	C7S	1.753 (6)
C1—C6	1.384 (6)	C8C9	1.491 (7)
C1—S	1.757 (4)	C8—N2	1.467 (6)
C2—C3	1.371 (5)	C10-C11	1.452 (10)
C3—C4	1.410 (5)	C10—N2	1.490 (7)
C3—N1	1.462 (5)	N1—O3	1.228 (5)
C4—C5	1.412 (6)	N1—04	1.216 (5)
C4N2	1.366 (5)	O1—S	1.430 (4)
C5—C6	1.364 (6)	O2—S	1.442 (3)
C4—C3—N1	122.0(3)	O3-N1-O4	122.8 (4)
C3—C4—C5	114.0 (4)	C8—N2—C4	123.4 (4)
C3C4N2	124.8 (4)	C10—N2—C4	119.8 (4)
O3—N1—C3	118.6 (4)	C8—N2—C10	116.1 (4)
O4-N1-C3	118.5 (3)		
C6-C1-S-O1	13.2(3)	C2C1SO2	-40.1(3)
C2C1SC7	74.2 (4)		

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989). Cell refinement: *CAD-4 Software*. Data reduction: *GX* (Mallinson & Muir, 1985). Program(s) used to solve structure: *MITHRIL* (Gilmore, 1984). Program(s) used to refine structure: *GX*. Molecular graphics: *ORTEPII* (Johnson, 1971; Mallinson & Muir, 1985). Software used to prepare material for publication: *GX*.

Lists of structure factors, anisotropic displacement parameters, Hatom coordinates and complete geometry have been deposited with the IUCr (Reference: HA1134). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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KR-25003, a Potent Analgesic Capsaicinoid

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Abstract

The crystal structure of N-[3-(3,4-dimethylphenyl)propyl](4-hydroxy-3-methoxyphenyl)acetamide, $C_{20}H_{25}NO_3$, was determined to a final R value of 0.047. The vanilloid, amide and dimethylphenyl groups of the compound are nearly perpendicular with respect to one another. The overall conformation in the crystalline state is somewhat different from that of other capsaicinoids. The molecules are stabilized by intermolecular hydrogen bonds.

Comment

Capsaicin, the pungent principle component of red pepper, is well known to have analgesic activity in animal species (Suzuki & Iwai, 1984). It has a different mechanism of action from the usual analgesics such as narcotic or non-steroidal anti-inflammatory drugs (NSAIDs). Although the undesirable side effects of natural capsaicin prevent its direct utilization in medicinal preparations for humans, the strong analgesic effect of the capsaicinoids, natural as well as synthetic, has been studied in detail (Dray, 1992).